

# Additions and Corrections

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**L. H. Zalkow, E. J. Eisenbraun, and J. N. Shoolery:** Structures of Carotol and Daucol.

Page 981. N.m.r. peak positions were given in this paper in dimensionless units defined as  $10.00 - 10^6(\nu - \nu_{\text{SiMe}_4})/\nu_{\text{SiMe}_4}$ . This definition has come to be generally designated by the symbol  $\tau$  rather than the symbol  $\delta$  as used in this paper, the latter usually being recognized as referring to peak positions defined as  $10^6(\nu - \nu_{\text{SiMe}_4})/\nu_{\text{SiMe}_4}$ . The data in this paper can be made more consistent with that appearing in other papers either by replacing  $\delta$  by  $\tau$  throughout, or by subtracting all chemical shift values from 10.00.

Page 982. The calculated value for the analysis of daucol acetate which appeared on this page should read as follows: "Anal. Calcd. for  $\text{C}_{17}\text{H}_{28}\text{O}_3$ : C, 72.82; H, 10.06; O, 17.12." instead of

"Anal. Calcd. for  $\text{C}_{17}\text{H}_{28}\text{O}_4$ : C, 73.34; H, 9.41; O, 17.24."

**Samuel C. Temin:** Polymers from Bisphenols. Steric Inhibition of Condensation Polymerization.

Page 2519. In Table I, for bisphenol diacetate no. 5,  $\text{R}_2$  should be " $\text{CH}_3$ " and not " $\text{C}(\text{CH}_3)_3$ ".

Page 2519. In Table I, the melting point of bisphenol diacetate no. 3 should be " $133\text{--}134^\circ\text{C}$ " and not " $59\text{--}60^\circ\text{C}$ ."

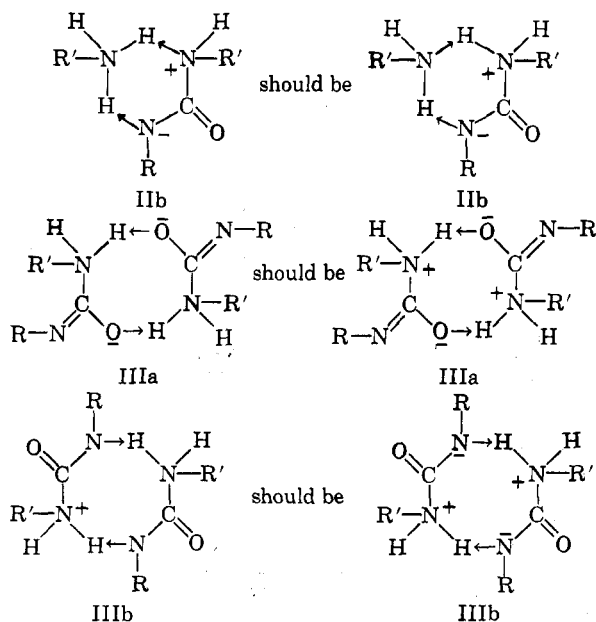
**Omar W. Steward and Ogden R. Pierce:** Fluoroalkyl and 3-(Fluoroalkoxy)propylpolysiloxanes.

Page 2944. In col. 2, lines 23 to 26, the absorption bands reported as C—F stretching bands probably should not be assigned to any localized vibration; however, they are characteristic of a 3,3,3-trifluoropropyl group attached to silicon. The assignment of a band to the C—Si bond is incorrect.

**Kenneth Sauer and Marie H. Kasparian:** Kinetics of the Reaction of Phenyl Isocyanate with Aniline.

Page 3502. In col. 2 the sentence following Equation 13 should read: A consequence of Equation 13 is the fact that for  $K_{-1} > k_2a$  the reaction should be third-order over all and for  $k_{-1} < k_2a$  it should be second-order over all.

Page 3503. In col. 2



**Konoshin Onodera and Tohru Komano:** Debenzyloxycarbonylation of 1,3,4,6-Tetra-*O*-acetyl-2-benzyloxycarbonyl-amino-2-deoxy-D-hexopyranoses in the Conversion of  $\alpha,\beta$ -Acetoxy to Glycosyl Bromide.

Page 3933. In col. 2, line 42, for "melting at  $115^\circ$ " read "melting at  $133^\circ$ ."

Page 3933. In col. 2, line 43 to line 44, for " $\text{C}_{22}\text{H}_{29}\text{O}_{10}\text{N}$ : C, 56.53; H, 6.21; N, 3.00. Found: C, 56.38; H, 6.27; N, 3.20" read " $\text{C}_{22}\text{H}_{27}\text{O}_{11}\text{N}$ : C, 54.86; H, 5.65; N, 2.91. Found: C, 54.72; H, 5.61; N, 2.93."

**Gordon N. Walker and Barbara N. Weaver:** Synthesis of Benzo[f]quinolines and Ergolines from 5-Phenyl-6-methyl-2-pyridones.

Page 4441. The structure of lysergic acid is depicted incorrectly, and should be altered by the insertion of  $-\text{NCH}_3$  in the appropriate position in Ring D.

Page 4444. Structure 26 should be altered by insertion of N in ring position 4.

Page 4448. In Experimental, compound "(II.  $\text{R} = \text{C}_6\text{H}_5$ )" is named incorrectly and should be 2,4-Diphenyl-3-methylpyrazole.

**Robert Barker and Hewitt G. Fletcher, Jr.:** 2,3,5-Tri-*O*-benzyl-D-ribose and -L-arabinose Bromides.

Page 4608. In col. 1, line 34, for "1,5-Anhydro-D-ribitol" read "1,4-Anhydro-D-ribitol."

**Elizabeth Dyer and Robert E. Read:** New Catalysts for the Conversion of Isocyanates to Carbodiimides.

Page 4678. In the last paragraph, 3rd line, delete "than". That is, the sentence should read: "... manganese naphthenate was the preferred catalyst due to the formation of fewer by-products with the naphthenate."

**H. M. Walborsky, L. Barash, and T. C. Davis:** Partial Asymmetric Syntheses: The Diels-Alder Reaction.

Page 4778. On lines 5, 10, 29, wherever dimethyl fumarate appears it should read *dimethyl* fumarate.

Since the publication of our results we discovered that A. Korolev and V. Mur, *Dokl. Akad. Nauk SSSR*, 59, 251 (1948) had reported obtaining a partial asymmetric synthesis in the Diels-Alder reaction.

**Harold E. Zaugg, Robert W. De Net, and Raymond J. Michaels:** Neighboring Group Reactions. II. A Novel Synthesis of Basic Esters of 1-Benzoxacycloalkanecarboxylic Acids.

Page 4832. In col. 2, line 5 from end, for "much more nucleophilicity" read "much lower nucleophilicity."

**R. K. Summerbell and Sherri R. Forrester:** A New Type of Assistance at a Distance.

Page 4837. In Table I, "1-Chloromethyl-4-iodomethyl-*p*-dioxane" should read "2-Chloromethyl-3-iodomethyl-*p*-dioxane".

**L. H. Sternbach and E. Reeder:** Quinazolines and 1,4-Benzodiazepines. IV. Transformations of 7-Chloro-2-methylamino-5-phenyl-3H-1,4-benzodiazepine 4-Oxide.

Page 4937. Formula Xa,b,d,e should have a single bond between 1 and 2 as shown below.

